## **Group Learning Activity - Chapter 2**

1. Draw line structures for two different compounds having the following characteristics.	
a) ether with the formula $C_4H_{10}O$	b) aromatic ester with the formula $C_9H_{10}O_2$
c) alkene with the formula $C_4H_8$	d) saturated aldehyde with the formula $C_6H_{12}O$
e) ketone with the formula $C_5H_{10}O$	f) anhydride with the formula $C_6H_{10}O_3$
g) amine with the formula $C_4H_9N$	h) a nitrile and alcohol with the formula $C_5H_9NO$
i) aryl halide with the formula $C_7H_7Br$	j) an alkyne and acid chloride with the formula $C_4H_3CIO$
k) carboxylic acid and nitro group with the formula $C_4H_7NO_4$	l) an amide which cannot form hydrogen bonds with the formula $C_5H_{11}NO$
m) unsaturated amine with the formula $C_4H_9N$	n) a saturated alkyl halide with the formula $C_5H_9I$