Homework - Chapter 15 Chem 2320

1. Write a reaction in which the following occurs. Briefly describe in what way the starting material is reacting (acid, base, nucleophile, electrophile, etc).
a) a tertiary alcohol reacts with sulfuric acid
b) the product of a) dissociates to form a carbocation
c) the carbocation in b) reacts with a water molecule to form an alkene
d) the carbocation in b) reacts with a chloride ion
e) a primary alcohol reacts with a carbocation
f) a primary alcohol reacts with phosphorus trichloride
g) a primary alcohol reacts with tosyl chloride in the presence of pyridine
h) the product of g) reacts with an acetylide anion

- i) a secondary alcohol reacts with tosyl chloride and pyridine
- j) the product of i) reacts as an acid with potassium tert-butoxide
- k) a secondary alcohol acts with sodium hydride
- 1) the product of k) reacts with a primary alkyl halide
- m) the product of k) reacts with the product of g)
- 2. Give the intermediate carbocation, then the products of the following reactions.

a)
$$H_2SO_4$$

e)
$$H_2SO_4$$

f)
$$\frac{\text{OH}}{\text{ZnCl}_2}$$

g)
$$H_3PO_4$$

h)
$$H_2SO_4$$
 NaBr

3. Give the mechanism for the following reactions, showing how all products are formed.

4. Give the products of the following reactions.

a)
$$\frac{\ddot{\text{OH}}}{\text{pyr}}$$

d)
$$PCl_5$$

5. Give the products of the following reactions. Draw arrows to show the reactions in the second step.

d)
$$\longrightarrow$$
 $\stackrel{\circ}{\text{OH}} \xrightarrow{\text{TsCl}}$ pyridine

e)
$$\frac{\text{TsCl}}{\text{pyridine}}$$
 Na^{+} $\frac{\text{-}}{\text{=}}$

6. Form the following ethers using an S_N2 reaction, using either an alkyl halide or a tosylate. Avoid side products from E2 reactions where possible; when side-products are unavoidable, make a note of it.

7. Give the products of the following reactions.

a)
$$Na_2CrO_4$$
 H_2SO_4, H_2O

c)
$$\stackrel{\text{OH}}{\longrightarrow} \frac{\text{K}_2\text{CrO}_4}{\text{H}_2\text{SO}_4,\text{H}_2\text{O}}$$

f)
$$CrO_3$$
 pyridine

$$g) \qquad \overbrace{\qquad \qquad } \stackrel{\text{OH}}{\longrightarrow} \qquad \underbrace{\qquad \qquad } \\ pyridine \qquad \qquad \\$$

8. Give the reagents that would be necessary to convert the following starting materials to the products shown.

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10. Give the compound that correspond to each letter in the following reaction sequence.

$$A \xrightarrow{PBr_3} B \xrightarrow{Mg} C$$

$$A \xrightarrow{H_2SO_4, H_2O} D$$

$$Na_2CrO_4$$

$$C + D \longrightarrow E \xrightarrow{H_3O^+} 3,4$$
-dimethyl-3-hexanol

$$A = D =$$

$$B = E =$$

$$C =$$

- 11. Write a reaction sequence (starting materials, reagents, and products) in which the following transformations occur.
 - a) primary alcohol --> alkyl chloride --> organolithium reagent --> alkane

b) secondary alcohol --> alkyl bromide --> Grignard reagent --> tertiary alcohol

d) ketone --> secondary alcohol --> alkoxide --> ether

e) alkene --> primary alcohol --> tosylate --> alkyne

f) primary alcohol --> aldehyde --> secondary alcohol

12. Fill in reagents and products which could be used to complete the following synthesis.